



**AN ALUM CATALYZED SOLVENT FREE ONE POT MULTICOMPONENT
SYNTHESIS OF 4-THIAZOLIDINONE DERIVATIVES**

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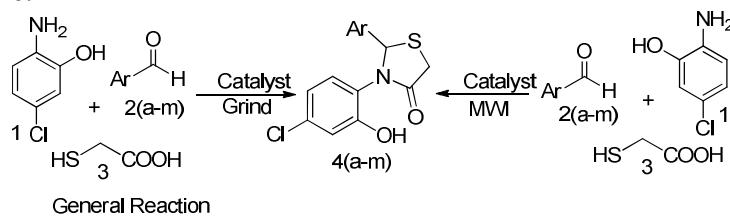
Abstract: 2-amino-5-chlorophenol (**1**) condensed with aromatic, heterocyclic-aldehydes (**2(a-m)**) and 2-mercaptopropanoic acid (**3**) in presence of alum catalyst, solvent free condition using grinding , microwave irradiation technique, synthesis of series of thiazolidinone derivatives (**4a-m**) expeditious in enviro-ecofriendly condition, with an excellent yield. The compounds are characterized by IR, NMR, CHN analysis.

Keywords: Thiazolidin-4-one, green approach, solvent free condition, alum catalyst.

Introduction

Heterocyclic compounds occur widely in nature and are essential to life. In particular five-membered heterocyclic molecules Sulfur and Nitrogen containing hetero atom having thiazole nucleus with carbonyl group on fourth carbon such as Thiazolidinone-4-one exhibit a variety of biological activities like anti-bacterialⁱ, anti-inflammatory^{ii,iii}, anticancer^{iv-vii}, anti-tuberculosis^{viii}, anticonvulsant^{ix} and analgesic^x. Previously studies on the pharmacological activities of thiazolidin 4-ones showed wide spectrums of antimicrobial activities^{xi-xiii}.

Alum [KAl(SO₄)₂.12H₂O] has been found to be effective in the synthesis of cis-isoquinolic acids^{xiv}, mono and disubstituted 2,3-dihydroquinazolin-4(1H)-ones^{xv}, dihydropyrimidines via Biginelli reaction^{xvi}, coumarins^{xvii}, 1,3,4-oxadiazoles^{xviii}, dibenzoxanthenes^{xix}, 1, 5 benzodiazepines^{xx}, and a different number of synthetic methods for 2-amino thiadiazole.^{xxi}



Many research study reported for green grinding method with formation of carbon carbon or carbon hetero atom single as well as double bond, some of name reactions example is Grignard^{xxii}, Reformatsky^{xxiii}, Dieckmann^{xxiv}, Aldolreaction^{xxv} and various other coupling type of reactions^{xxvi-xxviii}, an environmentally benign process are requested for crucial perventation of pollution, enviro-economic factors, low cost and simplicity in processing these factors are beneficial to industries as well as environment^{xxix}.

Microwave assisted reactions have become an established tool for the rapid synthesis of novel chemical entities.^{xxx} The application of microwave irradiation is the use of catalysts or mineral supported reagents, under solvent-free conditions in organic reactions to occur expeditiously at ambient pressure, thus providing exclusive chemical processes with special attributes such as enhanced reaction rates and higher yields^{xxxi-xxxii} and a functionalized thiadiazole

Experimental Section

2-amino-5-chlorophenol, substitute aldehydes, 2-mercaptoacetic acid (Thioglycolic acid) were commercially available. The major chemicals were purchased from Sigma Aldrich the progressed reaction monitored by TLC on silica gel precoated F254 Merc plates, the developed plates were examined with ultraviolet lamps (254nm) IR spectra were recorded on a FT-IR (Bruker). Melting points were recorded on open head capillary tube are uncorrected.¹H NMR spectra were recorded on a 400 MHz H¹ NMR spectra were recorded on a DRX-300 Bruker FT-NMR spectrophotometer The values of chemical shift are expressed in δ ppm as a unit.

Result and Discussion

Keeping focus on green protocol, we select alum as an efficient catalyst for the condensation of multicomponent in one pot synthesis. We used microwave irradiation and novel green approach techniques grinding. We herein report synthesis of a various thiazolidinone derivatives by novel, unique, expeditious method.

Initially, we optimized a different types of acid-catalyst (10 mol%) in appropriate time (Table 1.) among these alum is a very good catalyst for the increase the rate of reaction as well as yield of product. If we decrease amount of catalyst less than 10 mol % yield of product was fall down while increase amount of catalyst there is no significant effect (Table 1, entry no.7). Thus we choose 10 mol% alum for the synthesis of 4-thiazolidinones, all example were tested reasonably good to excellent yield (Table 2) could be achieved in short reaction time. An electronic effect was observed, electron with drawing and unsubstituted group gave better yield than electron donating substituent to the aromatic aldehydes (Table 2, entry no. 1,2,5), hetero-aryl aldehydes gave moderate yield (Table 2, entry no.12, 13). Finally, we observed an excellent yield, in short reaction time by MWI than grinding technique, first one is energy consume process while second one is totally green approach process, and structure of compound were substantiated by FTIR, ¹H NMR and CHN-analysis.

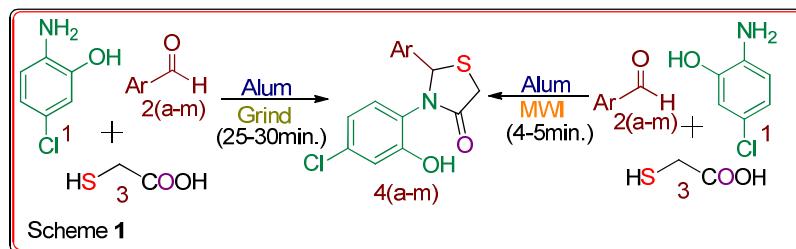
Table 1. Optimization of catalyst:

Sr. no.	Catalyst	Mole(%)	Isolated yield ^b (%)	
			Grind/Time(min.)	MWI/Time(min.)
1.	Mon.K-10	10	62/25	60/4
2.	PMA-SiO ₂	10	60/25	58/4
3.	AcOH	10	65/25	60/4
4.	Boric Acid	10	60/25	58/4
5.	Oxalic Acid	10	20/25	15/4
6.	Alum	10	93/25	98/4
7.	Alum	15	93/25	98/4

^bIsolated yield

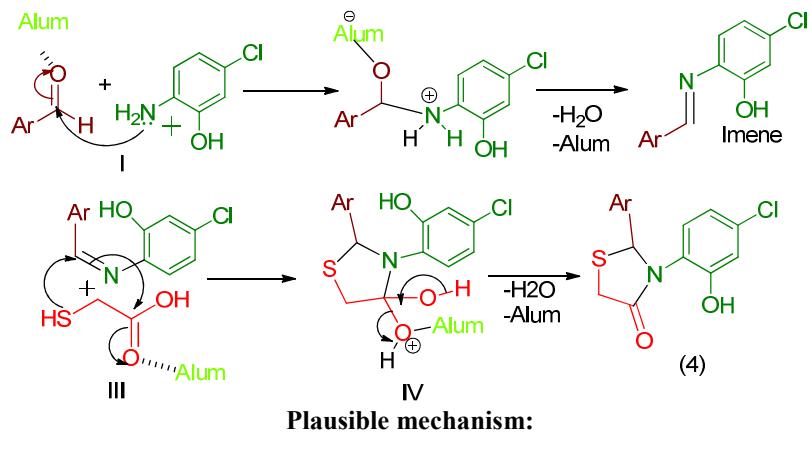
PMA= Phosphomolybdic acid supported on silica

Table 2. Synthesis of 4-thiazolidinones **4**

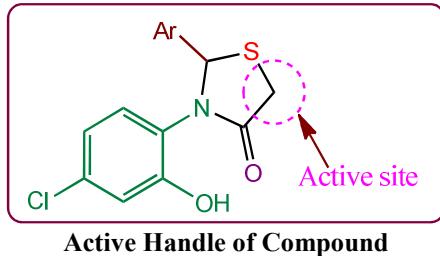


Sr.no.	Compoud	Ar	Yield 4 ^a (%) / Time (min.)		Melting Point (°C) (Lit.) ^{32(b)}	Color
			Grinding	MWI		
1	4a	H	93/25	98/4	89-91(88-90)	Yellow solid
2	4b	4-O ₂ NC ₆ H ₄	93/30	98/4	191-193(190-192)	Yellow solid
3	4c	2-ClC ₆ H ₄	91/30	96/5	128-130(127-129)	Yellow solid
4	4d	4-ClC ₆ H ₄	89/30	96/5	120(118-120)	Lightyellow solid
5	4e	2,4-Cl ₂ C ₆ H ₃	93/25	98/4	130-132(131-133)	Lightyellow solid
6	4f	4-FC ₆ H ₄	88/30	92/5	129(126-128)	Yellow solid
7	4g	3-FC ₆ H ₄	86/30	90/5	135-140(138-140)	Yellow solid
8	4h	4-H ₃ COC ₆ H ₄	89/30	92/5	51-53(48-50)	Yellow solid
9	4i	2,4(OCH ₃) ₂ C ₆ H ₃	88/30	93/5	45-50(46-48)	Yellow solid
10	4j	4-HOC ₆ H ₄	85/30	86/5	45(42-44)	Yellow solid
11	4k	4-benzonitrile	89/30	92/5	140-142(142-143)	Yellow solid
12	4l	4-thiazoyl	85/30	86/5	88-9290	Pale yellow solid
13	4m	2-Pyridyl	88/30	90/5	105-107(106-108)	Yellow solid

^a Isolated yield



Plausible mechanism:



General procedure For the synthesis of 4-thiazolidinone derivatives (4a-m):

Microwave irradiation method:

An equimolar amount of 2-amino-5-chlorophenol (1mmol) **1**, substitute aldehydes (1mmol) **2**, 2-mercaptopropanoic acid (Thioglycolic acid) (1mmol) **3** and Alum (10 mol%) were added in small RBF subjected to MW irradiation (800 watt.) at 100°C for 4-5 min. completion of reaction progress was monitored by TLC (Ethyl acetate :n-Hexane), the product obtained was poured in crushed ice, filtered and extracted with ethyl acetate and petroleum ether (8:2), compound were recrystallized from ethyl alcohol to obtained pure product (4a-m)

Grinding method:

An equimolar amount of 2-amino-5-chlorophenol (1mmol) **1**, substitute aldehydes (1mmol) **2**, 2-mercaptopropanoic acid (Thioglycolic acid) (1mmol) **3** and Alum (10 mol%) were taken in Mortar vigorous grind for 25-30 min. by Pestle at room temperature completion of reaction progress was monitored by TLC (Ethyl acetate :n-Hexane), Starting was disappear and the product obtained was poured in crushed ice, filtered and extracted with ethyl acetate and petroleum ether (8:2), compound were recrystallized from ethyl alcohol to obtained pure product,(4a-m), IR, NMR, CHN-Analysis of 3-(4-chloro-2-hydroxyphenyl) 2-phenylthiazolidin-4-one (**4a**):

IR (cm^{-1}): 3390, 3050, 2930, 1683

^1H NMR: δ ppm = 4.41–4.57 (d, 1H, CH_2), 4.60–4.69 (d, 1H, CH_2), 5.72 (s, 1H, OH), 6.23 (s, 1H, S–CH–N), 7.04–7.23 (m, 3H Ar-H), 7.21–7.27 (m, 5H, Ar-H)

^{13}C NMR: δ ppm = 33.2, 73.2, 116.8, 121.7, 124.2, 125.3, 126.2, 128.5, 129.1, 133.8, 139.4, 151.2, 171.9.; Anal. Calcd for $\text{C}_{15}\text{H}_{12}\text{ClNO}_2\text{S}$ (305.03): C, 58.92; H, 3.96; Cl, 11.51; N, 4.58; S, 10.47, Found: C, 58.93; H, 3.93; Cl, 11.55; N, 4.55; S, 10.45.

Spectral data:

3-(4-chloro-2-hydroxyphenyl) 2-phenylthiazolidin-4-one (4a):

IR (cm^{-1}): 3390, 3050, 2930, 1683

^1H NMR: δ ppm = 4.41–4.57 (d, 1H, CH_2), 4.60–4.69 (d, 1H, CH_2), 5.72 (s, 1H, OH), 6.23 (s, 1H, S–CH–N), 7.04–7.23 (m, 3H Ar-H), 7.21–7.27 (m, 5H, Ar-H)

^{13}NMR : δ ppm = 33.2, 73.2, 116.8, 121.7, 124.2, 125.3, 126.2, 128.5, 129.1, 133.8, 139.4, 151.2, 171.9.; Anal. Calcd for $\text{C}_{15}\text{H}_{12}\text{ClNO}_2\text{S}$ (305.03): C, 58.92; H, 3.96; Cl, 11.51; N, 4.58; S, 10.47, Found: C, 58.93; H, 3.93; Cl, 11.55; N, 4.55; S, 10.45.

3-(4-chloro-2-hydroxyphenyl) -2-(4-nitrophenyl)thiazolidin-4-one (4b):

IR (cm^{-1}): 3380 (OH), 3055 (Ar-H stretch.), 2935 (aliphatic CH stretch.), 1685 (C=O);

^1H NMR(δ ppm): 4.44.55 (d, 1H, CH_2), 4.57–4.65 (d, 1H, CH_2) 5.33 (s, 1H, OH), 6.60 (s, 1H, S–CH– N), 7.03–7.21 (m, 3H, Ar-H), 8.11–8.25(d,2H, Ar-H)

3-(4-chloro-2-hydroxyphenyl) -2-(2-chlorophenyl)thiazolidin-4-one (4c)

IR (cm^{-1}): 3390 (OH), 3061 (Ar-H stretch.), 2945 (aliphatic CH stretch.), 1690 (C=O); H

NMR: δ ppm = 3.99–4.09 (d, 1H, CH_2), 4.13–4.23 (d, 1H, CH_2) 5.20 (s, 1H, OH), 6.45 (s, 1H, S–CH– N), 7.02–7.23 (m, 3H, Ar-H), 7.24–7.37 (m, 4H, Ar-H).

3-(4-chloro-2-hydroxyphenyl) 2-(4-chlorophenyl)thiazolidin-4-one (4d):

IR (cm^{-1}): 3387 (OH), 3045 (Ar-H stretch.), 2935 (aliphatic CH stretch.), 1689 (C=O); 1H

NMR: δ ppm = 4.0 –4.15 (d, 1H, CH_2), 4.18–4.28 (d, 1H, CH_2), 5.31 (s, 1H, OH), 6.45 (s, 1H, S–CH– N), 7.05–7.27 (m, 3H,Ar-H), 7.27–7.37 (d, 2H, Ar H), 7.38–7.48 (d, 2H, Ar-H)

3-(4-chloro-2-hydroxyphenyl) 3-(4-chloro-2-hydroxyphenyl) (4e)

IR (cm^{-1}): 3393 (OH), 3051 (Ar-H stretch.), 2931 (aliphatic CH stretch.), 1685 (C=O); 1H

NMR: δ ppm = 4.46–4.56 (d, 1H, CH_2), 4.57–4.68 (d, 1H, CH 5.32 (s, 1H, OH), 6.41 (s, 1H, S–CH– N), 7.04–7.24 (m, 3H, Ar-H), 7.27–7.35(d, 2H,Ar-H)

3-(4-chloro-2-hydroxyphenyl)-2-(4-fluorophenyl)thiazolidin-4-one (4f):

IR (cm^{-1}): 3375 (OH), 3044 (Ar-H stretch.), 2934 (aliphatic CH stretch.), 1693 (C=O); 1H

NMR: δ ppm = 3.5–4.05 (d, 1H, CH_2), 4.08–4.18 (d, 1H, CH_2) 5.37 (s, 1H, OH), 6.47 (s, 1H, S–CH– N), 7.07–7.21 (m, 3H, Ar-H), 7.22–7.32 (d, 2H, Ar-H), 7.25–7.43 (d, 2H, Ar-H) Ar-H), 7.27–7.35 (d, 2H, ArH), 7.65 (s, 1H, Ar-H).

3-(4-chloro-2-hydroxyphenyl) -2-(3-fluorophenyl)thiazolidin-4-one (4g):

IR (cm^{-1}): 3388 (OH), 3039 (Ar-H stretch.), 2934 (aliphatic CH stretch.), 1687 (C=O); ^1H

NMR: δ ppm = 3.97–4.03(d1H, CH_2), 4.10–4.18 (d, 1H, CH_2) 5.30 (s, 1H, OH), 6.42 (s, 1H, S–CH–N), 6.77 (s, 1H, Ar-H), 7.06–7.21 (m, 3H, Ar-H), 7.22 –7.33 (m, 3H, Ar-H)

3-(4-chloro-2-hydroxyphenyl) -2-(4-methoxyphenyl)thiazolidin-4-one (4h):

IR (cm^{-1}): 3393 (OH), 3054(Ar-H stretch.), 2949 (aliphatic CH stretch.), 1686 (C=O);

^1H NMR: (δ ppm) 3.76 (s, 3H, CH_3), 3.90–3.98 (d, 1H, CH_2), 4.01– 4.12 (d, 1H, CH_2), 5.29 (s, 1H, OH), 6.44 (s, 1H (s, 1H, S–CH –N), 6.81–6.98 (d, 2H, Ar-H), 7.12–7.24 (m, 3H, Ar-H), 7.87–7.93 (d, 2H, Ar-H)

3-(4-chloro-2-hydroxyphenyl) 2-(2,4-dimethoxyphenyl)thiazolidin-4-one (4i):

IR (cm^{-1}): 3380 (OH), 3049 (Ar-H stretch.), 2939 (aliphatic CH stretch.), 1691 (C=O); ^1H NMR: δ ppm = 3.80 (s, 6H, CH_3), 3.89–4.08 (d, 1H, CH_2), 4.10–4.29 (d, 1H, CH_2), 5.32 (s, 1H, OH), 6.47 (s, 1H, S–CH–N), 6.57 (s, 1H, Ar-H), 7.02–7.34 (m, 3H, Ar-H), 7.24–7.33 (d, 2H, Ar-H)

3-(4-chloro-2-hydroxyphenyl) -2-(4-hydroxyphenyl)thiazolidin-4-one (4j):

IR (cm^{-1}): 3390 (OH), 3062 (Ar-H stretch.), 2933 (aliphatic CH stretch.), 1683 (C=O); ^1H NMR: δ ppm 3.97–4.01 (d, 1H, CH_2), 4.31–4.31 (d, 1H, CH_2), 5.55 (s, 2H, OH), 6.70 (s, 1H, S–CH–N), 6.73–6.68 (d, 2H, Ar-H), 7.02–7.21 (m, 3H, Ar-H), 7.757.01–7.20 (m, 3H, Ar-H), 7.73–7.88 (d, 2H, Ar-H).

4-(3-(4-chloro-2-hydroxyphenyl)-4-oxothiazolidin-2-yl)benzonitrile (4k):

IR (cm^{-1}): 3345 (OH), 3060 (Ar-H stretch.), 2931 (aliphatic CH stretch.), 2223 (CN), 1692 (C=O); ^1H NMR: δ ppm 3.70–3.81 (d, 1H, CH_2), 4.10–4.21 (d, 1H, CH_2), 5.36 (s, 1H, OH), 6.43 (s, 1H, S–CH–N), 7.31–7.30 (m, 3H, Ar-H), 7.40–7.45 (d, 2H, Ar-H), 7.51–7.67 (m, 2H, Ar-H).

3-(4-chloro-2-hydroxyphenyl)-2-(4-methylthiazol-5-yl)thiazolidin-4-one (4l):

IR (cm^{-1}): 3383 (OH), 3061(Ar-H stretch.), 2935 (aliphatic CH stretch.), 1689 (C=O); ^1H NMR: δ ppm = 2.41 (s, 3H, CH_3) 3.81–3.93 (d, 1H, CH_2), 4.01–4.21 (d, 1H, CH_2), 5.33 (s, 1H, OH), 5.81 (s, 1H, S–CH–N), 7.06,7.21 (m, 3H, Ar-H), 8.67 (s, 1H, Ar-H)

3-(4-chloro-2-hydroxyphenyl)2-(pyridin-2-yl)thiazolidin-4-one (4m):

IR (cm^{-1}): 3391 (OH), 3069(Ar-H stretch.), 2931 (aliphatic CH stretch.), 1685 (C=O); ^1H NMR: δ ppm = 91–3.91 (d, 1H, CH_2), 4.10–4.18 (d, 1H, CH_2), 5.38 (s, 1H, OH), 6.11 (s, 1H, S–CH–N), 7.01-7.20(m,3H,Ar-H), 7.32-7.43(m, 4H,Ar-H)

Conclusion

In conclusion we report the synthesis of 4-thiazolidinone derivatives in alum (10 mol%) catalyzed under solvent free condition using grinding and microwave assisted techniques, concluded that comparing MWI and Grinding techniques, expeditious with excellent yield was obtained by microwave irradiation method. Developed novel green grinding method as non requirement of heat and electrical energy.

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References:

- i. Chopra, I.; Schofield, C.; Everett, M.; O'Neill, A.; Miller, K.; Wilcox, M.; Frere, J. M.; Dawson, M.; Czaplewski, L.; Urleb, U.; Courvalin, P. *Lancet Infect. Dis*(**2008**), 8, 133
- ii. Uchoa, F. D. T.; Cattani, V. B.; Lima, M. C. A., Galdino, S. L ; Pitta, I. R.; Costa, T.D. *JBraz. Chem. Soc.* (**2008**), 191553.
- iii. Suthar, S. K.; Jaiswal, V.; Lohan, S.; Bansal, S.; Chaudhary, A.; Tiwari, A.; Alex, A. T.; Joesph, A. *Eur. J. Med. Chem*(**2013**), 63, 589.
- iv. Havrylyuk, D.; Mosula, L.; Zimenkovsky, B.; Vasylenko, O Gzella, A.; Lesyk R. *Eurr.Med. Chem.* (**2010**), 45, 5012.
V. Bhatt, J. J.; Shah, B. R.; Shah, H. P.; Trivedi, P. B.; Undavia, N. K.; Desai, N. C.*Indian J. Chem.*(**1994**), 33B, 189.
- vi. Patil, V.; Tilekar, K.; Mehendale -Munj, S.; Mohan, R.; Ramaa, C. S.*Eur. J. Med. Chem.* (**2010**), 45, 4539.
- vii. Chandrappa, S.; Kavitha, C. V.; Shahabuddin, M. S.; Vinaya, K.; Ananda, C. S.;Ranganatha, S. R.; Raghavan, S. C.; Rangappa, K. S.*Bioorg. Med. Chem.* (**2009**), 17, 2576.
- viii. Prasanna, P.; Balamurugan, K.; Perumal, S.; Yogeeswari, P.; Sriramb, D. *Eur. J. Med. Chem.*(**2010**), 45, 5653
- ix. Ragab, F. A.; Eid, N. M.; E El-Tawab, H.A. *Pharmazie.* (**1997**), 52, 926
- x. Knutsen, L. J. S.; C. Hobbs, J.; Earnshaw, C. G.; Fiumana, A.; Gilbert, J.; Mellor, S. L.; Radford, F.; Smith, N. J.; Birch, P. J.; Burley, J. R.;Ward S. D. C.; James, I. F.*Bioorg. Med. Chem. Lett.* (**2007**), 17,662.
- xi. El-Gaby, M. S. A.; El-Hag Ali, G. A. M.; El-Maghraby, A. A.; Abd El-Rahman, M. T.; Helal, M. H. M. *Eur. J. Med. Chem.* (**2009**), 44, 4148.
- xii. Vicini, P.; Geronikaki, A.; Incerti, M.; Zani, F.; Dearden, J.; Hewitt, M.*Bioorg. Med. Chem.* (**2008**), 16, 3714.
- xiii. Kavitha, C. V.; Basappa, Swamy, S. N.; Mantelingu, K.; Doreswamy, S.; Sridhar, M. A.;Prasadb, J. S.; Rangappa, K. S. *Bioorg. Med. Chem.* (**2006**), 14, 2290
- xiv. J. Azizian, A.A. Mohammadi, A.R. Karimi and M.R. Mohammadizadeh, *J. Org. Chem.* (**2005**), 70, 350–352.
- xv. M. Dabiri, P. Salehi, S. Otokesh, M. Baghbanzadeh, G. Kozehgary and A.A. Mohammadi,*Tetrahedron Lett.* (**2005**), 46, 6123–6126.
- xvi. J. Azizian, A. A. Mohammadi, A. R. Karimi and M. R. Mohammadizadeh, *Appl.CatalA. General* (**2006**), 300, 85–88.
- xvii. M. Dabiri, M. Baghbanzadeh, S. Kianiand Y. Vakilzadeh, *Monatsh. Chem.* (**2007**), 138, 997–999.
- xviii. M. Dabiri, M. Baghbanzadeh, and M. Bahramnejad, *Monatsh. Chem.*, (**2007**), 138,1253–1255.
- xix. M. Dabiri, M. Baghbanzadeh, M.S. Nikchehand E.Arzroomchilar, *Bioorg. Med. Chem. Lett.* (**2008**), 18, 436–438.
- xx. D. Mahajan, T. Nagvi, R.L. Sharma, K.K. Kapoor, *Aust. J. Chem.*(**2008**), 61,159–162.
- xxi. Santosh A. Jadhav, Pardeshi R. K.,Shioorkar M. G., Chavan O. S. Scholars Research Library *Der Pharma Chemica*, (**2015**), 7 (2)127-131
- xxii. Toda F.; Takumi H.;Yamaguchi H.;*Chem.Exp.* (**1989**), 4, 507.
- xxiii. Tanaka K.; Kishigami S.; Toda F.; *J.Org. Chem.* (**1991**), 56, 4333.
- xxiv. Toda F.; Suzuki T.; Higa S.; *J. Chem. Soc. Perkin trans.* (**1998**), 1, 3521.

- xxv. Toda F.; Tanaka Hummai.; *J. Chem. Soc. Perkin trans.* (1990), 1, 3203.
- xxvi. Toda F.; Tanaka K.; Iwata. S.; *J. Org. Chem.* (1989), 54, 3007.
- xxvii. Toda F.; Kiyoshige K.; Yagi. M.; *AngewanteChem*, Int. Ed. (1989), 2, 320
- xxviii. (a) Schmeyers T.; Toda F.; Boys J. Kaupp. G. *J. Chem Soc. Perkin trans.* (1998), 2, 989.; (b) Xiao J. P.; Wang Y. L.; Zau. Q. X.; *Synth. Commun.* (2001), 31, 66.; (c) Wang Z. X.; Hua, L. Q.; *Green Chem.* (2004), 60, 90.
- xxix. Tanaka K.; Toda F.; *Chem. Rev.* (2000), 100, 1025.
- xxx. Bolognese, A.; Correale, G.; Manfra, M.; Lavecchia, A.; Novellino, E.; Barone, V. *Biomol. Chem.* (2004), 2, 2809.
- xxxi. Varma, R. S. *Green Chem.*(1999), 1, 43.
- xxxii. (a)Larhed, M.; Hallberg, A.*Drug Disc. Today* (2001), 6(8), 406.(b) R. N. Shelke, D. B. Shinde, *J. Med. Chemistry and Drug Discovery*.ISSN:2347-9027, ACTRA 2015.

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